SYNTHESIS AND PROPERTIES OF 16-HYDROXY ANALOGS OF PGE₂ M Bruhn, C H. Brown, P W Collins, J R Palmer, E Z Dajani and R Pappo* Searle Laboratories P O Box 5110 Chicago, Illinois 60680

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As an extension of our work^{1,2} on the total synthesis of PGE_1 and its analogs we have investigated the synthesis of the corresponding compounds in the PGE_2 series for evaluation as inhibitors of gastric secretion

Taking advantage of our experience on the conjugate addition of 1-alkynes to α , β -unsaturated ketones,¹ we prepared the required keto acid <u>6</u> by a novel approach. Starting from commercially available 5-chloro-1-pentyne <u>1</u>, the corresponding lithium salt was obtained using butyl lithium at -40° in toluene. This acetylide was then converted <u>in situ</u> to its trialkynyl boron derivative <u>2</u> by treatment with boron trichloride or boron trifluoride etherate at -40° first and then overnight at -10°. The resulting mixture was cooled to -20° and after addition of methyl vinyl ketone, the 1,4-addition reaction was allowed to proceed at -40° for 4 hours. The adduct <u>3</u> was thus obtained in 48% yield after acid hydrolysis, b p 76° (0 09 mm), PMR ¹¹ (s) $\delta 2$ 12, (t) $\delta 3$ 61

It is noteworthy that the substitution of 2 by the corresponding dimethyl alkynylalane^{3,4} in this reaction gave the 1,4 adduct $\underline{3}$, accompanied by an equal amount of the 1,2 adduct

Refluxing 3 with sodium cyanide in aqueous ethanol gave the nitrile 4, IR (CHCl₃) 2250 cm⁻¹ and 1715 cm⁻¹, PMR (s) δ 2 15, which was hydrolyzed readily to the acid 5. Hydrogenation of 5 with palladium on barium sulfate yielded the desired <u>cis</u>-olefinic acid 6, PMR (s) δ 2 12, (t) δ 5 38, (s) δ 10 2

$$HC = C(CH_2)_{3}C1 \xrightarrow{1} \underbrace{n-BuL_1}_{2 \quad BF_3 \quad OEt_2} B(C = CCH_2CH_2CH_2C1)_{3} \xrightarrow{1} \underbrace{CH_3CCH=CH_2}_{2 \quad H_3O^+} CH_3C(CH_2)_{2}C = C(CH_2)_{3}C1$$

$$\xrightarrow{1} CH_3C(CH_2)_{2}CH = CH(CH_2)_{3}COOH \leftarrow CH_3C(CH_2)_{2}C = C(CH_2)_{3}COOH \leftarrow CH_3C(CH_2)_{2}C = C(CH_2)_{3}CN$$

$$\xrightarrow{6} \xrightarrow{5} \underbrace{4}$$

The conversion of the keto-acid <u>6</u> to the key cyclopentenone derivative <u>13</u> was accomplished using the procedures developed previously in our laboratories for the synthesis of PGE₁ derivatives ⁵ Thus <u>6</u> was condensed with dimethyloxalate in the presence of potassium <u>t</u>-butoxide to yield the glyoxalate <u>7</u> The crude reaction product was then treated with aqueous hydrochloric acid at reflux and the resulting crude triketo acid <u>8</u> was purified either by chromatography or via the corresponding monooxime <u>9</u>, m p 163-164°, PMR (D₂O) (t) δ 5 47, (p) δ 1 70, (s) δ 3 13, (d) δ 2 98, (t) δ 2 43, (q) δ 2 22, to give crystalline <u>8</u>, m p 84-85°, PMR (s) δ 2 95, (t) δ 5 50, (d) \circ 3 17, (dd) δ 1 55- δ 2 O2

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Treatment of <u>8</u> with aqueous sodium borohydride at $0-5^{\circ}$ gave the hydroxydione <u>10</u>, m p 83-84°, PMR (D₂0) (t) δ 5 45, (s) δ 4.75, (dd) δ 4 63, (d) δ 2 85, (dd) δ 2 94, (dd) δ 2 32 Alternatively the monooxime <u>9</u> was reduced with titanium trichloride in aqueous acetic acid buffered with excess sodium acetate to yield <u>10</u> in 77% yield, the intermediate trione <u>8</u> produced being reduced further by the reagent

The ketol <u>10</u> was converted to the enol ether <u>11</u>, m p 77-78°, PMR (dd) δ 3 12, (s) δ 3 68, (s) δ 3 98, (m) δ 4 28, (m) δ 5 39, advantage being taken of its low solubility in ether relative to its oily isomer <u>12</u>, as described previously for the PGE₁ series ⁵ The ketone group of the enol ether <u>11</u> was reduced with sodium dihydro-bis(2-methoxyethoxy)aluminate at $-\delta$ 5° in toluene and the resulting alcohol <u>14</u> rearranged and hydrolyzed <u>in situ</u> with dilute acid to <u>13</u>, which was purified by chromatography over Mallinckrodt Silicar CC7 silica gel to yield an oil, PMR (dd) δ 2 27, (dd) δ 2 87, (s) δ 3 68, (m) δ 4 96, (m) δ 7 20, (t) δ 5 55 This compound <u>13</u> prepared by a different route, has been previously described ⁶



The key intermediate <u>13</u> was then separated into its enantiomers using the method described previously for the resolution of methyl 3-hydroxy-5-oxocyclopent-1-eneheptanoate ⁷ Treatment of <u>13</u> with (R)-2-aminooxy-4-methylvaleric acid <u>15</u> yielded a mixture of oximes which were separated by chromatography using a mixture of 0.5% acetic acid and 1.5% isopropanol in chloroform as eluent on E. Merck Silica Gel to yield <u>16</u>, $[\alpha]_D^{\approx}$ -53.4 (CHCl₃), and <u>17</u>, $[\alpha]_D^{\approx}$ +46.3 (CHCl₃) ⁸ Regeneration of the ketones <u>18</u>, $[\alpha]_D^{\approx}$ -9.9 (MeOH), and <u>19</u>, $[\alpha]_D^{\approx}$ +8.4 (MeOH), was accomplished with titanium trichloride ⁷

The configuration at C-3 of <u>18</u> and <u>19</u> was determined by comparing the ORD and CD curves with the published data⁷ on the analogous (3R)- and (3S)-methyl-3-hydroxy-5-oxocyclopent-1-ene-1-heptanoate Proof of structure of <u>13</u> was obtained by converting it to $(\pm)PGE_2$ methyl ester via the cuprate reaction described previously ²

The conversion of <u>13</u> and <u>10</u> to potent gastric anti-secretory compounds was then undertaken. Thus 1,4 addition of the cuprate reagent derived from <u>trans</u>-1-10do-4,4-dimethyl-1octen-3-ol 3-<u>t</u>-butyldimethylsilyl ether⁹ to the tetrahydropyranyl ether derivative of <u>19</u> (<u>20</u>) led to a mixture of products which was separated by chromatography on silica gel to yield 16,16-dimethyl PGE₂ methyl ester, ¹⁰ PMR (s) 60 86, (s) 60 90, (s) 63 69, (d) 63 86, (q) 64 10,





(m) $\delta 5 \ 45-5 \ 93$, (m) $\delta 5 \ 39-5 \ 45$, $J_{14,15} \ 6.5$, $J_{13,14} \ 15 \ 0$, and $15-ep_1-16,16-dimethyl \ PGE_2 \ methyl \ ester$, PMR (s) $\delta 0 \ 88$, (s) $\delta 0 \ 91$, (s) $\delta 3 \ 68$, (d) $\delta 3 \ 88$, (q) $\delta 4 \ 12$, (m) $\delta 5 \ 45-5 \ 93$, (m) $\delta 5 \ 30-5 \ 45$, $J_{14,15} \ 5 \ 5$, $J_{13,14} \ 15 \ 0$



Interestingly the use of the cuprate reagents $\underline{21}$ and $\underline{22}^2$ led to a new and very important series of prostaglandin analogs. Thus reaction of the tetrahydropyranyl ether of $\underline{13}$ with $\underline{21}$ led to $(\pm)15$ -deoxy-16-hydroxy PGE₂ methyl ester ($\underline{23}$), PMR (p) §1 67, (dd) §2 75, (s) §3 69, (q) §4 08, a mixture of two racemates isomeric at C₁₆, which was active as a gastric anti-secretory agent. Similarly $\underline{22}$ afforded $(\pm)15$ -deoxy-16-hydroxy-16-methyl PGE₂ methyl ester ($\underline{24}$), PMR (s) §1 19, (p) §1 68, (dd) §2 75, (q) §4 09, again as a mixture of two racemates, isomeric at C₁₆ Moreover the reaction of the tetrahydropyranyl derivative of the (3R)-ketone $\underline{19}$ with the racemic cuprate reagent $\underline{22}$ yielded a mixture of (16R)-15-deoxy-16-hydroxy-16-methyl PGE₂ methyl ester and (16S)-15-deoxy-16-hydroxy-16-methyl PGE₂ methyl ester $\left[\alpha\right]_{D}^{250}$ = -47 7 (MeOH) The separation of these isomers is under study

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In line with the results disclosed previously for the E_1 analogs,² the 16-hydroxy derivative 24 was found to be an extremely potent gastric anti-secretory agent when administered either intravenously or intragastrically to Heidenhain pouch dogs

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- 8 Using (S)2-aminooxy-4-methylvaleric acid gave similar results except that the order of elution of the oximes on chromatography was reversed.
- 9 This reagent was prepared by the general procedure described in reference 2 from 4,4dimethyl-1-octyn-3-ol by addition of catechol borane to the corresponding 3-t-butyldimethylsilyl ether, followed by iodination to yield the trans-1-iodo-4,4-dimethyl-1-octen-3-ol silyl ether Halogen metal interconversion proceeded rapidly with butyl lithium leading to the cuprate derivative after treatment with CuC=C(CH₂)₂CH₃
- 10 B J Magerlein, D W DuCharme, W E Magee, W. L Miller, A Robert and J R Weeks, Prostaglandins, 4, 143 (1973)
- 11 All PMR spectra were run in deuteriochloroform unless noted otherwise