

SYNTHESIS AND PROPERTIES OF 16-HYDROXY ANALOGS OF PGE₂

M Bruhn, C H. Brown, P W Collins, J R Palmer,
E Z Dajani and R Pappo*
Searle Laboratories
P O Box 5110
Chicago, Illinois 60680

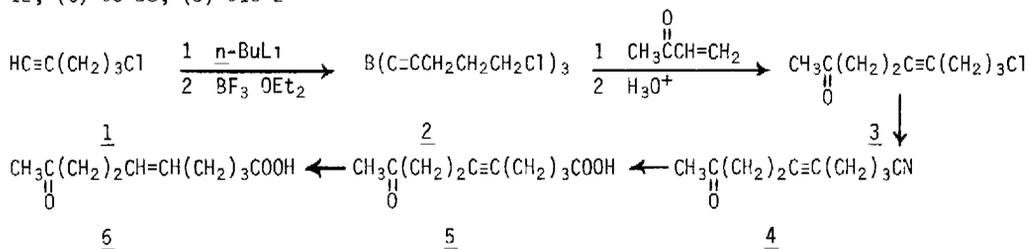
(Received in USA 19 November 1975; received in UK for publication 16 December 1975)

As an extension of our work^{1,2} on the total synthesis of PGE₁ and its analogs we have investigated the synthesis of the corresponding compounds in the PGE₂ series for evaluation as inhibitors of gastric secretion

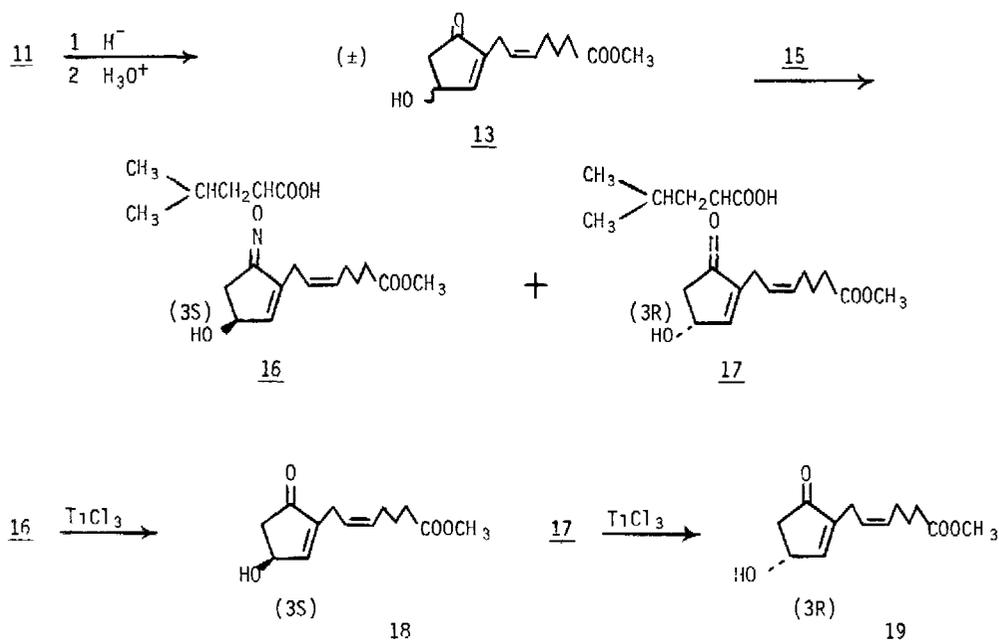
Taking advantage of our experience on the conjugate addition of 1-alkynes to α,β -unsaturated ketones,¹ we prepared the required keto acid 6 by a novel approach. Starting from commercially available 5-chloro-1-pentyne 1, the corresponding lithium salt was obtained using butyl lithium at -40° in toluene. This acetylide was then converted *in situ* to its trialkynyl boron derivative 2 by treatment with boron trichloride or boron trifluoride etherate at -40° first and then overnight at -10°. The resulting mixture was cooled to -20° and after addition of methyl vinyl ketone, the 1,4-addition reaction was allowed to proceed at -40° for 4 hours. The adduct 3 was thus obtained in 48% yield after acid hydrolysis, b p 76° (0.09 mm), PMR ¹¹ (s) δ 2.12, (t) δ 3.61.

It is noteworthy that the substitution of 2 by the corresponding dimethyl alkynylalane^{3,4} in this reaction gave the 1,4 adduct 3, accompanied by an equal amount of the 1,2 adduct.

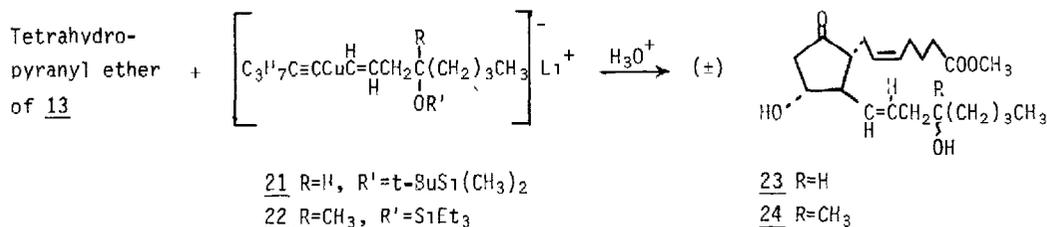
Refluxing 3 with sodium cyanide in aqueous ethanol gave the nitrile 4, IR (CHCl₃) 2250 cm⁻¹ and 1715 cm⁻¹, PMR (s) δ 2.15, which was hydrolyzed readily to the acid 5. Hydrogenation of 5 with palladium on barium sulfate yielded the desired *cis*-olefinic acid 6, PMR (s) δ 2.12, (t) δ 3.38, (s) δ 10.2.



The conversion of the keto-acid 6 to the key cyclopentenone derivative 13 was accomplished using the procedures developed previously in our laboratories for the synthesis of PGE₁ derivatives.⁵ Thus 6 was condensed with dimethyl oxalate in the presence of potassium *t*-butoxide to yield the glyoxalate 7. The crude reaction product was then treated with aqueous hydrochloric acid at reflux and the resulting crude triketo acid 8 was purified either by chromatography or via the corresponding monooxime 9, m p 163-164°, PMR (D₂O) (t) δ 5.47, (p) δ 1.70, (s) δ 3.13, (d) δ 2.98, (t) δ 2.43, (q) δ 2.22, to give crystalline 8, m p 84-85°, PMR (s) δ 2.95, (t) δ 5.50, (d) δ 3.17, (dd) δ 1.55- δ 2.02.



(m) δ 45-5 93, (m) δ 30-5 45, $J_{14,15}$ 6.5, $J_{13,14}$ 15 0, and 15-epi-16,16-dimethyl PGE₂ methyl ester, PMR (s) δ 0 88, (s) δ 0 91, (s) δ 3 68, (d) δ 3 88, (q) δ 4 12, (m) δ 5 45-5 93, (m) δ 5 30-5 45, $J_{14,15}$ 5 5, $J_{13,14}$ 15 0



Interestingly the use of the cuprate reagents 21 and 22² led to a new and very important series of prostaglandin analogs. Thus reaction of the tetrahydropyranyl ether of 13 with 21 led to (\pm)-15-deoxy-16-hydroxy PGE₂ methyl ester (23), PMR (p) δ 1 67, (dd) δ 2 75, (s) δ 3 69, (q) δ 4 08, a mixture of two racemates isomeric at C₁₆, which was active as a gastric anti-secretory agent. Similarly 22 afforded (\pm)-15-deoxy-16-hydroxy-16-methyl PGE₂ methyl ester (24), PMR (s) δ 1 19, (p) δ 1 68, (dd) δ 2 75, (q) δ 4 09, again as a mixture of two racemates, isomeric at C₁₆. Moreover the reaction of the tetrahydropyranyl derivative of the (3R)-ketone 19 with the racemic cuprate reagent 22 yielded a mixture of (16R)-15-deoxy-16-hydroxy-16-methyl PGE₂ methyl ester and (16S)-15-deoxy-16-hydroxy-16-methyl PGE₂ methyl ester [α]_D^{25°} = -47.7 (MeOH). The separation of these isomers is under study.

In line with the results disclosed previously for the E₁ analogs,² the 16-hydroxy derivative 24 was found to be an extremely potent gastric anti-secretory agent when administered either intravenously or intragastrically to Heidenhain pouch dogs

Acknowledgments We thank Professor H C Brown for very stimulating discussions. The authors are pleased to acknowledge the assistance of the following persons: Dr C Ens, Mr C C Kim and Mr C E Scott for the preparation of intermediates, Dr R H Bible, Jr, Ms L Swenton and Ms P M Green for interpretation of NMR data, Mr B G Smith and the chromatography department for chromatographic separations

REFERENCES

1. R Pappo, P Collins, Tetrahedron Letters, 2627 (1972)
2. P W Collins, E Z Dajani, M S Bruhn, C H Brown, J R Palmer and R Pappo, Tetrahedron Letters, in press
3. J Hooz and R B Layton, J Am Chem Soc, 93, 7320 (1971)
4. J Hooz and R Layton, Can J Chem, 50, 1105 (1972)
5. R Pappo, P Collins, C Jung, Ann. N. Y. Acad Sci, 64 (1971)
6. C J Sih, J. B Heather, R Sood, P Price, G Peruzzoti, L F H Lee, and S S Lee, J. Am. Chem. Soc., 97, 865 (1975)
7. R Pappo, P Collins and C Jung, Tetrahedron Letters, 943 (1973)
8. Using (S)-2-aminooxy-4-methylvaleric acid gave similar results except that the order of elution of the oximes on chromatography was reversed.
9. This reagent was prepared by the general procedure described in reference 2 from 4,4-dimethyl-1-octyn-3-ol by addition of catechol borane to the corresponding 3-*t*-butyldimethylsilyl ether, followed by iodination to yield the *trans*-1-iodo-4,4-dimethyl-1-octen-3-ol silyl ether. Halogen metal interconversion proceeded rapidly with butyl lithium leading to the cuprate derivative after treatment with $CuC\equiv C(CH_2)_2CH_3$
10. B J Magerlein, D W DuCharme, W E Magee, W. L Miller, A Robert and J R Weeks, Prostaglandins, 4, 143 (1973)
11. All PMR spectra were run in deuteriochloroform unless noted otherwise